

Quantum Brownian motion

and its conflict with the second law

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Extraction of work from a single thermal bath in the quantum regime, Phys. Rev. Lett. 85 (2000) 1799

Breakdown of the Landauer bound for information erasure in the quantum regime,
Phys. Rev. E 64, 056117 (2001)

Statistical thermodynamics of quantum Brownian motion: Construction of perpetuum mobile of the second kind, Phys. Rev. E, in press

On testing the violation of the Clausius inequality in nanoscale electric circuits
Phys. Rev. B, in press

Setup

- Harmonic quantum particle
in harmonic oscillator bath \rightarrow exactly solvable
- Thermodynamic interpretation
- Energy relaxation
- Entropy production
- Energy dispersion
- Work extraction cycles
- Summary

Quantum Brownian motion

$$\mathcal{H}_{tot} = \frac{p^2}{2m} + \frac{1}{2}ax^2 + \sum_i \left[\frac{p_i^2}{2m_i} + \frac{m_i\omega_i^2}{2} \left(x_i - \frac{c_i x}{m_i\omega_i^2} \right)^2 \right]$$

particle + oscillator bath + interaction $x_i x + x^2$

- $\omega_i = i \Delta$: level spacing $\Delta \rightarrow 0$:
- Spectral density: $J(\omega) = \sum_i \frac{\pi c_i^2}{2m_i\omega_i} \delta(\omega_i - \omega)$
- Ohmic: $J(\omega) = \gamma\omega$
- quasi-Ohmic: $J(\omega) = \frac{\Gamma^2}{\omega^2 + \Gamma^2} \gamma\omega$
(Drude-Ullersma)

“Caldeira-Leggett” model; Ullersma 1966

Applications

- fluctuation effects in Josephson junctions
- low-temperature quantum transport
- quantum-optical systems

At $t = 0$:

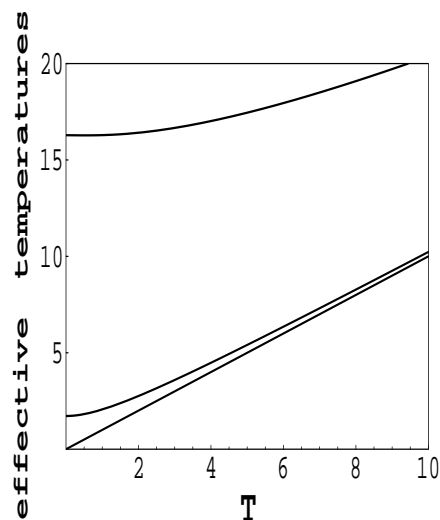
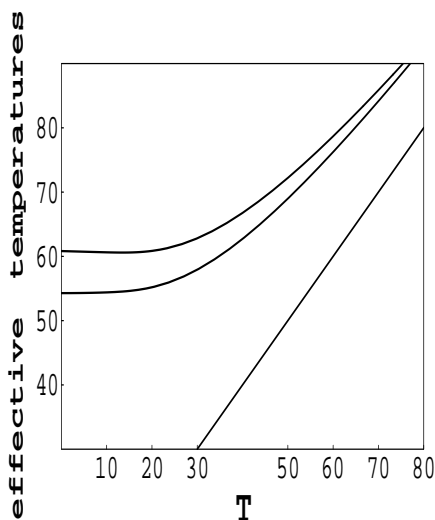
- Density matrix of total system is Gibbsian at T
- Spring constant $a_0 \rightarrow a = (1 + \alpha)a_0$ ($|\alpha| \ll 1$)
- **Work** done in switching: $\mathcal{W}_0 = \frac{1}{2}(a - a_0)\langle x^2 \rangle_0$
- $t > 0$: Subsystem (Brownian particle) relaxes

Stationary distribution: Wigner function
(Haake, Reibold 1985; Unruh, Zurek 1989):

$$W_{st} = W_p(p) \times W_x(x) = \frac{1}{Z_p} \exp\left(-\frac{p^2}{2mT_p}\right) \times \frac{1}{Z_x} \exp\left(-\frac{ax^2}{2T_x}\right)$$

It is **non-Gibbsian** since $T_x \neq T_p$ (both exceeding T)

→ equilibrium thermodynamics **endangered**



Thermodynamic description for adiabatic changes

$$U = \langle \mathcal{H} \rangle = \int dp dx \mathcal{H}(p, x) W(p, x) \equiv \int \mathcal{H} W$$

0) First law: $dU = \int \mathcal{H} dW + \int W d\mathcal{H} \equiv \delta Q + \delta W$

1) Violation of Clausius inequality $\delta Q \leq T dS$

$$\delta Q(T \rightarrow 0) = \frac{\hbar \gamma}{2\pi m^2} dm \neq 0, \quad \text{comes from cloud}$$

2) One can adsorb heat, $\delta Q > 0$, and become more localized, $dS < 0$.

Breakdown of Landauer bound for erasure of one bit of information $|\delta Q| \geq T \ln 2$

3) Generalized Clausius inequality:

$$\delta Q = T_p dS_p + T_x dS_x \quad \text{as for glasses \& black holes}$$

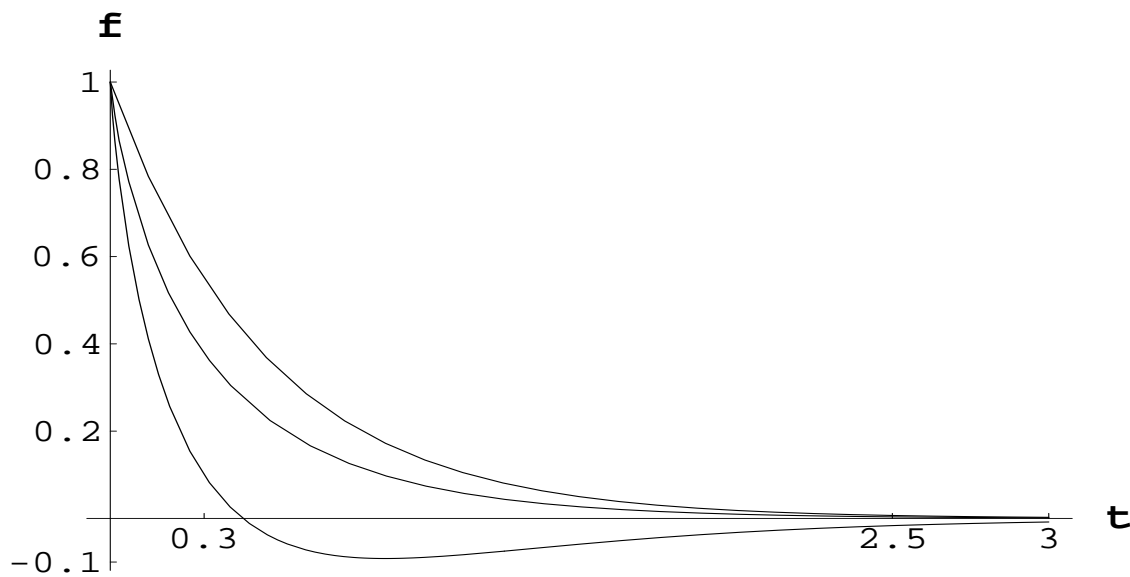
Boltzmann entropies of p and x - parts of Wigner function

$$S_p = - \int dp W_p(p) \ln W_p(p), \quad S_x = - \int dx W_x(x) \ln W_x(x)$$

Relaxation of the energy

$$U(t) = \frac{1}{2}T_x + \frac{1}{2}T_p + \alpha C(t), \quad \alpha = \frac{a-a_0}{a}$$

- Strong damping: ($\gamma \gg \sqrt{am}$)



Upper: large T ;

Middle: moderate T

Lower: $T = 0$; the integral vanishes

Production of Boltzmann entropy

$$S_B = - \int dx dp W(x, p, t) \ln [\hbar W(x, p, t)]$$

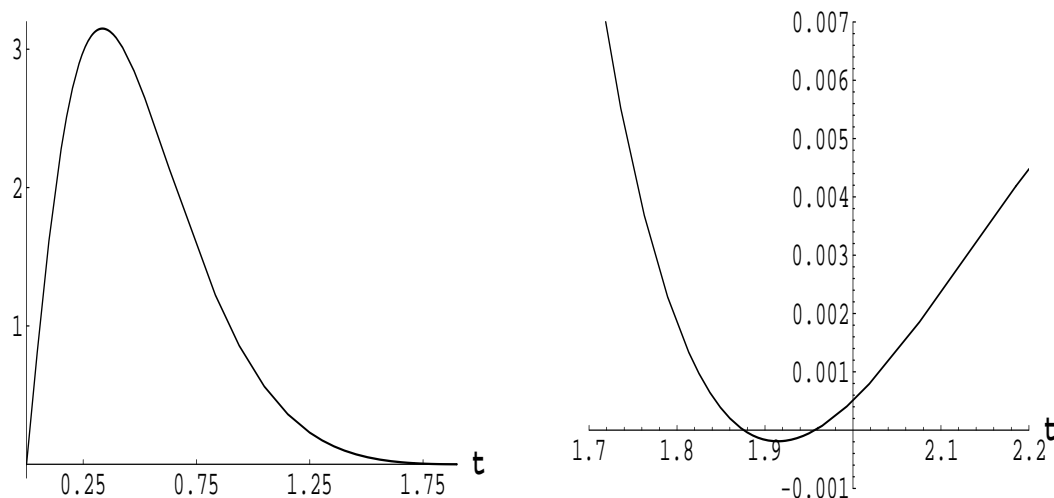
$$\text{change} = \text{flow} + \text{production}: \quad \frac{dS_B}{dt} = \frac{d_e S_B}{dt} + \frac{d_i S_B}{dt}$$

$$\text{entropy flow: } \frac{d_e S_B}{dt} = \frac{\partial_t \langle ax^2 \rangle / 2}{D_x(t)} + \frac{\partial_t \langle p^2 \rangle / 2m}{D_p(t)}$$

$D_x \neq D_p \neq T$: Diffusion coefficients from Fokker-Planck equation for Wigner function

Rate of entropy production: $\frac{d_i S_B}{dt}$ quadratic in α

- Strong damping, $T = 0$



Entropy production as function of time

Rate of energy dispersion:

$$\mathring{d}\mathcal{W} = \mathring{d}\mathcal{W}_{\text{adiabatic}} + \mathring{d}\Pi$$

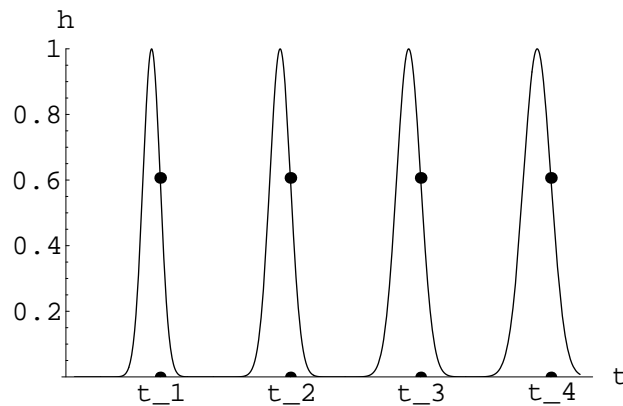
Slow change of $a(t) = [1 + \alpha(t)]a$ at small T

$$\frac{\mathring{d}\Pi}{dt} = T^2 \dot{\alpha}^2 + \dot{\alpha}\ddot{\alpha} - \dot{\alpha}\partial_t^3 \alpha$$

- Full change: Thomson formulation of second law

$$\Delta\Pi = \int_{-\infty}^{\infty} dt \frac{\mathring{d}\Pi}{dt} = \int dt [T^2 \dot{\alpha}^2 + \ddot{\alpha}^2] \geq 0$$

- Fixed time: $\dot{\alpha}\ddot{\alpha}$ term dominates: $\frac{\mathring{d}\Pi}{dt} < 0$ possible



Many $\sim \frac{1}{T}$ work extraction cycles possible.

What goes wrong with thermodynamics ?

- The interaction energy is non-zero: $U_{int} \sim \gamma$
Thermodynamics needs not apply when
 $T < T_* \sim \gamma$
- But rate of entropy production negative (and energy relaxation non-monotonous) for $T \gg T_*$
- Rate of energy dispersion (of total system) can be negative when $T < T_*$
- Out of equilibrium $\sim \frac{1}{T}$ cycles possible in which energy is extracted.
“Perpetuum mobile of second kind”
Optimal extraction if one cycle